Mössbauer and optical absorption studies of aegirines

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Two chemically well analyzed natural aegirines were studied by Mössbauer and optical absorption spectroscopy. Aegirines belong to the clinopyroxenes and have the space group C2/c. The pure ferric is called compound is called acmite and has the chemical formula NaFe³⁺Si₂O₆. In the acmite structure Fe sites are occupies the sixfold coordinated M1 sites and Na⁺ fills the eightfold coordinated M2 sites. Acmite has the typical yellow-greenish colour originated by Fe³⁺ in silicates. In agreement with this, the optical absorption spectrum of one of our samples consists mainly of absorption bands due to Fe³⁺ in a slightly distorted octahedral ligand-field, probably intensified by Fe³⁺- Fe³⁺ interactions mainly in the α -spectrum. The Mössbauer spectrum of the same acmite shows a ferric doublet, which is assigned to Fe at the M1 site.

In the second aegirine sample some Fe^{3+} is replaced by Fe^{2+} and the crystals are now dark-green and strongly pleochroic. The optical absorption spectrum of this sample reveals some Fe^{2+} absorption bands in addition to Fe^{3+} in the near infrared spectral region. The distinct pleochroism is mainly caused by a Fe^{2+} - Fe^{3+} intervalence charge transfer band at 13330 cm⁻¹, which is strongly polarized in the α -spectrum, e.g. almost parallel to the direction of the c-axis. The Mössbauer spectrum of the second sample exhibits, in addition to an Fe^{3+} doublet, an Fe^{2+} doublet both assigned to M1. There is another broad intermediate absorption pattern of which relative area increases on raising temperature in the Mössbauer spectrum, which represents some Fe^{2+} and Fe^{3+} ions involved in a thermally activated electron delocalization process. The present study suggests that electron delocalization in aegirines occurs preferentially within the infinite chains formed by edge-sharing M1 octahedra parallel to the c-axis.